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(May)

PHYSICS

(Major)

Course : 602

(Condensed Matter Physics)

Full Marks : 60

Pass Marks : 24/18

Time : 3 hours

*The figures in the margin indicate full marks
for the questions*

1. Answer the following as directed : $1 \times 6 = 6$

(a) A metal crystallizes with a face-centred cubic lattice. The edge of the unit cell is 408 pm. The diameter of the metal atom is

~~(i)~~ 204 pm

(ii) 288 pm

(iii) 408 pm

~~(iv)~~ 144 pm

(Choose the correct answer)

(b) What is the coordination number of the HCP structure?

(c) The effective number of free electrons in a completely filled band is zero.

(State True or False)

(d) Write the relation between average kinetic energy of electrons in the ground state (\bar{E}_0) with Fermi energy in one-dimensional crystal.

(e) The slope of the $\ln \sigma$ (σ is conductivity) versus $1/T$ plot is a measure of

(i) mobility

(ii) resistivity

(iii) band gap

(iv) None of the above

(Choose the correct answer)

(f) The Meissner effect in superconductor is a/an

(i) reversible process

(ii) irreversible process

(iii) isothermal process

(iv) adiabatic process

(Choose the correct answer)

2. (a) Calculate the efficiency of packing in the case of a metal crystal in simple cubic lattice.

2

(b) How energy levels of an atom become energy band in a solid?

2

(c) The intrinsic carrier concentration in an Si sample is 1.5×10^{16} atoms/m³. It is doped with 10^{23} phosphorus atoms/m³. Determine its hole concentration and conductivity. Given electron mobility = $0.135 \text{ m}^2 \text{ V}^{-1} \text{ s}^{-1}$.

2

3. (a) Calculate the binding energy of an ionic crystal and obtain an expression for the Madelung constant. Evaluate Madelung constant for a linear ionic crystal.

3+1+2=6

(b) Find the angle between [111] and [110] directions of a cubic lattice.

2

4. (a) Describe the nature and origin of various forces existing between the atoms of a crystal. Explain the formation of stable bond using the potential energy versus interatomic distance curve.

3+3=6

Or

How does Bragg's reflection differ from ordinary reflection? What is Ewald construction? How does it help to interpret X-ray diffraction photographs?

1+2+3=6

(b) Prove that f.c.c. lattice is reciprocal to b.c.c. lattice.

2

5. (a) Write down the postulates of free electron gas model. A particle of mass m is confined in a field-free region between impermeable walls at $x=0$ and $x=L$. Show that the stationary energy levels of the particles are given by

$$E_n = \frac{n^2 h^2}{8mL^2}$$

2+5=7

Or

What is density of states? Show that the density of states at the Fermi surface is

$$D(E_F) = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2} \right)^{3/2} E_F^{1/2}$$

1+6=7

(b) Find the relation between Fermi energy and average kinetic energy of an electron at absolute zero temperature.

3

6. (a) State and explain the Bloch theorem. Discuss its importance in the band theory. 2+3=5

(b) What is the nature of potential experienced by an electron in a crystal? Using the Kronig-Penney model, show that for $p \ll 1$, the energy of the lowest energy band is

$$E = \frac{\hbar^2 p}{ma^2} \quad 1+4=5$$

7. (a) What is meant by Fermi level? Sketch the Fermi level in p -type and n -type semiconductors. Show that the Fermi level of an intrinsic semiconductor lies at the middle of the band gap. 1+2+4=7

(b) State two basic characteristics of superconductors. Explain the difference between type-I and type-II superconductors using Meissner effect. 1+4=5

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